

**Quantitative analysis of eicosanoids and other oxylipins –
Investigation of oxidative stress and inflammation by means of targeted
metabolomics of oxylipins in cell culture**

Katharina M. Rund and Nils Helge Schebb

Chair of Food Chemistry, Faculty of Mathematics and Natural Sciences, University of
Wuppertal, Gaußstr. 20, 42119 Wuppertal

Katharina M. Rund: E-Mail: katharina.rund@schebb-web.de
Nils Helge Schebb: E-mail: nils@schebb-web.de, Tel: +49 202-439-3457;

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1 What you will learn in this chapter

- 2 • Application of a quantitative targeted metabolomics approach to lipid mediators as an
3 important complement to non-targeted metabolomics and semiquantitative methods
4 described in this book
- 5 • Concept of quantitative LC-MS analysis using external calibration and internal standards
- 6 • Strategies for the analysis of low abundant endogenous compounds/lipids
- 7 • Basic cell culture strategies
- 8 • Evaluation of oxidative stress based on lipid peroxidation products
- 9 • Key lipid mediators formed during inflammation in the cyclooxygenase and lipoxygenase
10 pathways of the arachidonic acid cascade and their pharmacological modulation
- 11 • Lipid mediators are functional markers with dual function reflecting both non-enzymatic
12 formation by autoxidation and enzymatic formation as cellular response

14 Abstract

15 Eicosanoids and other oxylipins are oxygenation products from polyunsaturated fatty acids
16 (PUFA). They can be formed non-enzymatically by (lipid) autoxidation as well as by specific
17 enzymatic reactions. In mammals, enzymatic formation comprises three main pathways catalyzed
18 by cyclooxygenases (COX), lipoxygenases (LOX) and cytochrome P450 monooxygenases (CYP).
19 Several of the oxylipins are highly potent lipid mediators, playing a key role in the regulation of
20 pain, fever and inflammation. Thus, oxylipin formation is a major drug target. In fact, common
21 pharmaceuticals being sold without prescription, nonsteroidal anti-inflammatory drugs (NSAID),
22 directly target COX.

23 In this chapter, a detailed protocol is described to investigate oxylipin formation during autoxidation
24 and upon an inflammatory stimulus in cell culture. The low abundant lipid mediators are quantified
25 by means of targeted liquid chromatography-tandem mass spectrometry (LC-MS/MS). The
26 experiments demonstrate that oxylipins are markers of cell biology, reflecting the redox
27 status/oxidative stress on the one hand and the cellular inflammatory response on the other.

28 Based on a selected set of oxylipins the students learn how to set up a quantitative LC-MS/MS
29 method and apply it to biological samples. With the detailed protocol, cell culture experiments can
30 be rapidly set up in the laboratory to investigate the effect of oxidative stress or inflammatory
31 stimuli in cells. The strategy also allows to test new compounds for their efficacy to reduce
32 oxidative stress or to alleviate proinflammatory lipid mediator formation. Thus, the described
33 procedures facilitate the implementation of individual research projects in advanced practical
34 student courses.

35 Overall, the experiments lead to a comprehensive understanding of quantitative LC-MS/MS as
36 well as basic cell culture experiments to study the bioactivity of new compounds. The investigation
37 and interpretation of the results enables to understand oxylipin formation and their role in biology.

38 Keywords

39 oxidative stress, inflammation, oxylipins, eicosanoids, LC-MS, quantitative analysis, cell culture,
40 arachidonic acid cascade, leukotrienes, prostaglandins, isoprostanes

41 Abbreviations

4-HNE	4-hydroxy-2-nonenal	LT	leukotriene
ARA	arachidonic acid (C20:4 n6)	LTA ₄	5 <i>S-trans</i> -5,6-oxido-7 <i>E</i> ,9 <i>E</i> ,11 <i>Z</i> ,14 <i>Z</i> -eicosatetraenoic acid
BCA	bicinchoninic acid	LTAH	leukotriene A4 hydrolase
BHT	butylated hydroxytoluene	LTB ₄	5 <i>S</i> ,12 <i>R</i> ,6 <i>Z</i> ,8 <i>E</i> ,10 <i>E</i> ,14 <i>Z</i> -dihydroxy-eicosatetraenoic acid
CE	collision energy	<i>m/z</i>	mass to charge ratio
COX	cyclooxygenase	MDA	malondialdehyde
CXP	collision exit potential	MeOH	methanol
CYP	cytochrome P450 monooxygenase	MRM	multiple reaction monitoring
DHA	docosahexaenoic acid (C22:6 n3)	MS	mass spectrometry
DiHETE	dihydroxy eicosatetraenoic acid	MS/MS	tandem mass spectrometry
DiHETrE	dihydroxy eicosatrienoic acid	n3-PUFA	omega-3 polyunsaturated fatty acid
DMEM	Dulbecco's modified Eagle medium	n6-PUFA	omega-6 polyunsaturated fatty acid
DMSO	dimethylsulfoxide	NSAID	non-steroidal anti-inflammatory drug
DP	declustering potential	PBS	phosphate buffered saline
EDTA	ethylenediaminetetraacetic acid	PCR	polymerase chain reaction
EP	entrance potential	PG	prostaglandin
EPA	eicosapentaenoic acid (C20:5 n3)	PGF _{2α}	9α,11α,15 <i>S</i> -trihydroxy-prosta-5 <i>Z</i> ,13 <i>E</i> -dien-1-oic acid
EpETrE	epoxy-eicosatrienoic acid	PUFA	polyunsaturated fatty acid
ESI	electrospray ionization	ROOH	organic hydroperoxide
EtOH	ethanol	RP	reversed phase
FCS	fetal calf serum	rpm	rounds per minute
GIT	gastrointestinal tract	RPMI	Rosewell Park Memorial Institute
HAc	acetic acid	RSD	relative standard deviation
HETE	hydroxy-eicosatetraenoic acid	S/N	signal-to-noise ratio
HHT/HHTrE	hydroxy heptadecatrienoic acid	SD	standard deviation
HpETE	hydroperoxyeicosatetraenoic acid; hydroperoxy-ARA	SPE	solid phase extraction
IC50	inhibitory concentration 50%	SRM	selected reaction monitoring
IS	internal standard	<i>t</i> -AUCB	<i>trans</i> -4-[4-(3-adamantan-1-yl-ureido)- cyclohexyloxy]-benzoic acid
IsoP	isoprostane	<i>t</i> -BOOH	<i>tert</i> -butyl hydroperoxide
KOH	potassium hydroxide	TGF-β1	transforming growth factor-β1
LC	liquid chromatography	<i>t</i> _R	retention time
LLOQ	lower limit of quantification	TX	thromboxane
LOD	limit of detection	ULOQ	upper limit of quantification
LOX	lipoxygenase	UPLC	ultra high performance liquid chromatography
LPS	lipopolysaccharide		

42

43

44 1 Introduction

45 Polyunsaturated fatty acids (PUFA) occur in all biological systems and are major constituents of
46 the polar lipids in the cell membrane. These PUFA act as precursors of signaling molecules –
47 following a concept widely used by biological systems: The generation of signaling molecules from
48 an abundant species of a major class of biomolecules such as, e.g. amino acids (serotonin,
49 epinephrine, histamine, thyroxine, etc.) or cholesterol (cortisol, testosterone, estradiol, vitamin D,
50 etc.). The signaling molecules resulting from PUFA – lipid mediators – comprise a multitude of
51 different oxygenated PUFA. A challenge for analytical methods is that the concentration of the
52 signaling molecules is low, and orders of magnitude lower than those of their precursors. Targeted
53 analysis of lipid mediators is currently carried out by LC-MS/MS as summarized in (1, 2) allowing
54 the selective and sensitive detection and quantification of multiple analytes in parallel.

55 Lipid mediators are formed from PUFA by oxidation leading to eicosanoids (oxygenated C20
56 PUFA) and other oxylipins (all oxygenated PUFA). Because arachidonic acid (ARA, C20:4, n6) is
57 the major n6-PUFA occurring in mammalian cells, it plays a key role in the formation of highly
58 biologically active lipid mediators (3). Though this chapter focusses on ARA-derived eicosanoids

59 (which are best investigated), it should be noted that oxylipins can be formed from all PUFA. This
60 is why the diet, i.e. the intake of long-chain n3-PUFA, such as EPA and DHA, directly modulates
61 the oxylipin formation and pattern in the organism (4).

62 PUFA bearing one or more 1,5 pentadiene systems (two double bounds with a methylene group
63 (CH₂) in between) can easily be chemically oxidized by free radicals and singlet oxygen (5, 6).
64 Initially hydroperoxides (ROOH) are formed reacting further to a plethora of oxidation products. A
65 break in the carbon chain leads to aldehydes, which are the cause of rancidity in food (plant oils,
66 potato chips, etc.) or reactive markers (e.g. malondialdehyde (MDA), 4-hydroxy-2-nonenal (4-
67 HNE)) of oxidative stress. This oxidative stress is characterized by a misbalance of radical
68 production and removal processes in cells and associated with several diseases (7-9) – even
69 though it is often unclear if it is its cause or the consequence. In order to investigate oxidative
70 stress in cells, tissues and the whole organism stable products are needed.

71 Oxylipins can serve here as ideal biomarkers: Reduction of initially formed hydroperoxides
72 (ROOH, Hp) – which takes place rapidly in the cell through the action of glutathione reductase –
73 leads to hydroxy-PUFA, in case of ARA particularly 5-, 9-, 11- and 15-hydroxy-eicosatetraenoic
74 acid (HETE) (Fig. 1), whereof those which are not enzymatically formed can be used as marker
75 of oxidative stress. More specifically, cyclic stable reaction products formed *via*
76 bicycloendoperoxide intermediates from the initial peroxy-fatty acids – so-called isoprostanes
77 (IsoP) – are established biomarkers for oxidative stress, induced e.g. by smoking, cardiovascular
78 or neurological diseases (10, 11).

79 In the first experiment of this chapter, oxidative stress is induced in a cell line by incubation with
80 *tert*-butyl hydroperoxide (*t*-BOOH), directly inducing lipid peroxidation. This leads to a time- and
81 dose-dependent increase of isoprostanes demonstrating its applicability to monitor oxidative
82 stress in biological systems. Analysis by LC-MS/MS using the targeted oxylipin metabolomics
83 approach additionally enables the simultaneous quantification of hydroxy-PUFA, i.e. HETEs
84 derived from hydroperoxy-PUFA (HpETE), which can be formed by autoxidation as well as by
85 enzymes (see below). Moreover, the effect of oxidative stress on epoxy-PUFA, i.e. EpETrE is
86 explored. Though epoxidation of double bounds is not a dominating reaction in the course of
87 autoxidation, we recently uncovered that the ratio of the stereoisomers (*cis*- and *trans*-epoxy-
88 PUFA) is a new marker of oxidative stress (12). Based on the evaluation of these three classes of
89 oxylipins the power of quantitative oxylipin metabolomics to assess oxidative stress is
90 demonstrated and enables to learn about the biology of (non-enzymatic) oxylipin formation by
91 correlating the different results.

92 Conversion of PUFA by the enzymes of the ARA-cascade leads to similar oxygenation products.
93 However – as for all enzymatic catalysis – the reactions are much more selective and give rise to
94 specific products (3). Though some of these products can also be formed by non-enzymatic
95 reactions, the product resulting from enzymatic conversion is just one of a large number of
96 possible regio- and stereoisomers (e.g. prostaglandin F_{2α} (PGF_{2α})) or even not formed without a
97 specific enzyme which is highlighted in the second experiment for leukotriene B₄ (LTB₄).

98 Three (super)families of enzymes catalyze the formation of eicosanoids and other oxylipins. By
99 far the best investigated is the cyclooxygenase (COX) pathway. COX generates an unstable

100 bicycloendoperoxide-peroxy product (PGG₂) from ARA, which is reduced to the unstable PGH₂
101 which can be further converted by specific synthases (and in case of PGE₂, PGD₂ and TXB₂ also
102 non-enzymatically) to prostaglandins (PG), thromboxane (TX) and prostacyclin (10, 13). The latter
103 are regulators of, e.g. pain, inflammation, fever and blood coagulation. Thus, COX is a major target
104 of widely used pharmaceuticals involving the best-selling over-the-counter drugs aspirin,
105 diclofenac and ibuprofen, summarized as non-steroidal anti-inflammatory drugs (NSAIDs). Two
106 isoforms of COX exist: COX-1 being constitutively expressed, e.g. in the gastrointestinal tract
107 (GIT), and the inducible COX-2 playing a key role in inflammation. In the second experiment a
108 monocytic cell line (THP-1), which is differentiated to a macrophage-like phenotype is used. Here,
109 an inflammatory response, and thus induction of COX is elicited by lipopolysaccharide (LPS), a
110 constituent of the cell wall of Gram-negative bacteria. This is assessed based on a dramatic
111 increase in the production of PGE₂, a major prostanoid formed during acute inflammation, as well
112 as two non-enzymatically formed PGH₂ breakdown products: 12-HHTrE and TXB₂ (14). The
113 experimental setting also allows the testing of common drugs, enabling to investigate hands-on
114 the mechanism and biological effect of drugs known from daily life.

115 Different hydroxy-PUFA are formed (through reduction of initially generated hydroperoxides) as
116 side products of COX activity, i.e. 11-HETE and 15 HETE (15, 16). Again, it is a key learning for
117 the interpretation of data from metabolomics to evaluate and interpret the levels of these oxylipins,
118 which can be formed *via* different enzymatic as well as non-enzymatic pathways. A highly
119 interesting extension of the experiment could be the investigation of aspirin, irreversibly blocking
120 prostanoid formation, while COX-2 is still giving rise to 15-H(p)ETE (15, 17, 18).

121 The second pathway of the ARA-cascade is catalyzed by several lipoxygenases, forming in the
122 first step positional and stereospecific hydroperoxides, which can be reduced to hydroxy-PUFA,
123 HETE in the case of ARA. The nomenclature of the enzymes originates from the position where
124 the hydro(peroxy) group is inserted within the ARA molecule. Two 12-LOX, two 15-LOX and the
125 5-LOX are found in humans (19).

126 In the second experiment only the 5-LOX pathway is investigated giving rise to physiologically
127 active leukotrienes: The initially formed 5-hydroperoxy fatty acid is converted by 5-LOX to the
128 unstable epoxide LTA₄ (Fig. 1). Similar to PGH₂ in the prostanoid formation pathway, this
129 intermediate is then transformed by specific enzymes. The LTC₄ synthase reduces the epoxy
130 group forming a glutathione conjugate, which causes muscle contraction of the smooth muscles
131 playing a key role in the regulation of the lung function (not shown). Hydrolysis by leukotriene A₄
132 hydrolase (LTAH) leads to LTB₄ a highly potent chemoattractant for neutrophils. This means, if
133 this compound is released by a given cell, it attracts neutrophils to come to the site increasing the
134 (local) inflammatory response. In this experiment, the formation of LTB₄ is analyzed in THP-1 cells.
135 Moreover, in parallel also the isomers of LTB₄ (*trans*- and *epi*-isomers of LTB₄ as well as 5,6-
136 DiHETE isomers) are monitored which are chemical breakdown products of the unstable LTA₄
137 formed when the short-lived (about 20 s in a biological setting) LTA₄ is not timely converted by
138 LTAH.

139 The third pathway of the ARA-cascade is catalyzed by cytochrome P450 monooxygenase (CYP)
140 enzymes giving also rise to highly potent lipid mediators (20, 21). Terminal hydroxylation results
141 in formation of 20-HETE – a hydroxy-PUFA that cannot be generated by autoxidation – playing a

142 key role in the regulation of blood pressure. Moreover, epoxy-PUFA (Fig. 1) are formed acting as
143 potent anti-inflammatory and vasodilatory mediators. Investigating the formation of these
144 mediators is beyond the scope of the current educational experiments (for an example for the
145 investigation of the modulation of the CYP pathway by phytochemicals see (22)).

146 It should be noted that the experiments carried out in this chapter only highlight a few of the large
147 number of oxylipins formed from ARA and all other PUFAs, which can be in parallel investigated
148 using the described methodology (for the methods please refer to (14, 23-25)).

149 With the application of quantitative targeted oxylipin metabolomics for the evaluation of oxidative
150 stress, the experiment demonstrates how oxylipins can be used as markers for this
151 (patho)physiological condition.

152 In the second set of experiments, oxylipins are monitored as active mediators of inflammation,
153 generated by COX-2 and 5-LOX. It is demonstrated how these oxylipins can be modulated by
154 stimuli and how their formation can be blocked by drugs. Here, targeted metabolomics serves as
155 indispensable tool to monitor the target engagement of the drugs.

156

157 **Chemicals and materials**

158 All chemicals and materials needed for the experiments are listed the supplementary material
159 (Table S1). Preparation of solutions for cell culture, SPE and LC-MS/MS analysis is described in
160 the supplementary material.

161 **Step-by-step protocol starts**

162 **2 Step-by-step protocol**

163 **2.1 Cell cultivation: Maintaining cells in culture**

164 *Thawing of cells stored in cryo tubes in liquid nitrogen*

- 165 a. Gently thaw the cell pellet in the cryo tube in a water bath (37 °C, approx. 2 min)
- 166 b. Transfer the cells under sterile conditions to a 15 mL conical centrifuge tube containing
167 10 mL warmed medium (see sections 2.2 and 2.3 for medium composition)
- 168 c. Centrifuge (5 min, room temperature, 600 × g)
- 169 d. Remove the medium
- 170 e. Add 7 mL (for HCT-116) or 5 mL (for THP-1) fresh medium and resuspend the cells
171 thoroughly
- 172 f. Seed/transfer the cells:
 - 173 i. For HCT-116 (adherent): Seed in a T25 flask (25 cm²) for adherent cells
 - 174 ii. For THP-1 (suspension cells): Seed in a 22.1 cm² dish for suspension cells

175 g. Check growth and morphology of the cells under the microscope the following days,
176 replace the medium with fresh if needed and transfer cells to bigger dishes when almost
177 confluent

178 *NOTE: Use the cells only for at most 10 passages after defrosting, enabling the replication of*
179 *experiments from frozen cell stocks (at comparable passage number).*

180 *Check absence of mycoplasmas (every second month):* These are small bacteria which can
181 contaminate the cell culture and alter the biology of the test system. To verify their absence in the
182 cell culture, collect an aliquot of the cell culture medium from nearly confluent cells (right before
183 transfer) and determine mycoplasmas using, e.g. a commercial mycoplasma detection kit for
184 conventional PCR.

185 *Generation of a lab stock of frozen cells*

- 186 a. Transfer cells from the dish to a conical centrifuge tube
 - 187 i. For HCT-116 (adherent cells): Detach cells from the dish using trypsin (as described
188 in section: 2.2 “Cell culture: oxidative stress”, step 3. a. – g)
 - 189 ii. For THP-1 (suspension cells): Gently resuspend cells before transfer
- 190 b. Determine the cell number in the suspension, e.g. with a Neubauer chamber
- 191 c. Centrifuge (5 min, room temperature, 600 × g)
- 192 d. Remove the medium and resuspend the cells in cold fetal calf serum (FCS) containing
193 10% DMSO leading to 2 × 10⁶ cells/mL (HCT-116) or 10 × 10⁶ cells/mL (THP-1)
- 194 e. Transfer 1 mL in cryo tubes and freeze at -80 °C overnight using a cell freezing
195 container (e. g. “Mr. Frosty”) allowing a freezing rate of 1 °C/min which is the optimal
196 condition to conserve cells
- 197 f. Transfer the cryo tubes to liquid nitrogen storage

198 *NOTE: Check the viability of the cells in an aliquot of the generated frozen cell pellet by*
199 *performing the thawing steps described under section 2.1 a – g.*

200 **2.2 Cell culture assay: Oxidative Stress in HCT-116 cells**

- 201 1. The adherent human colorectal carcinoma cell line HCT-116 (obtained from the German
202 Collection of Microorganisms and Cell Cultures GmbH (DSMZ, Braunschweig, Germany))
203 is cultivated in Dulbecco's modified Eagle medium (DMEM) supplemented with 10% FCS,
204 100 U/mL penicillin, 100 µg/mL streptomycin (P/S, 2%) and 2 mM L-glutamine (1%) in a
205 humidified incubator at 37 °C and 5% CO₂ in 10 cm dishes (60.1 cm²) for adherent cells

206 *Maintaining the cell culture:* Transfer cells every 2-3 days

- 207 2. Replace medium every 2 days (if no cell passaging is carried out, otherwise see step 3.)
208 a. Carefully remove medium (e.g. using a membrane pump and pipette)
209 b. Replace with fresh warmed medium
- 210 3. Passaging every 2-3 days (Monday, Wednesday, Friday)
211 a. Carefully remove medium
212 b. To detach the cells, add 1.5 mL trypsin (0.25% trypsin in PBS-EDTA) to the cells,
213 swivel the dish to wet the surface completely and remove the trypsin directly
214 c. Add 1.5 mL trypsin (0.25% trypsin in PBS-EDTA), let it act while swiveling the dish
215 for 30 s and remove again
216 d. Incubate the cells on the dish without liquid for 1.5 min at 37 °C
217 e. Tap the dish e.g. against the palm of the hand or the bench and observe if the cells
218 are detaching and moving from the surface of the dish. If the cells do not move,
219 incubate for another 10 s at 37 °C and check again. Repeat in steps of 10 s until
220 the cells are detaching
- 221 *NOTE: If protein expression levels are intended to be determined the use of trypsin for*
222 *detachment of the cells should be omitted to avoid premature protein digestion. For*
223 *this, scraping is recommended to detach the cells from the surface of the dish (see*
224 *section: 2.3 "Cell culture: inflammatory response", step 4. a).*
- 225 f. Add 10 mL fresh warmed medium and flush the surface of the dish thoroughly to
226 detach and remove all cells from the surface of the dish
227 g. Transfer the cells and medium to a conical centrifuge tube
228 h. Determine the cell number
229 i. Seed 1×10^6 (on Monday and Wednesday) or 0.8×10^6 (on Friday) cells in a
230 total of 10 mL fresh medium in a new dish 10 cm (60.1 cm²) for adherent cells
- 231 4. Incubation with *tert*-butyl hydroperoxide (*t*-BOOH)
232 a. Seed 2×10^6 cells in a total of 10 mL fresh medium in a new dish 10 cm (60.1
233 cm²) for adherent cells
234 b. After 24 h of growth remove the medium, add fresh medium (without FCS) and
235 add 10 µL *t*-BOOH (in water). For dose- and time-dependent investigation of the
236 effects recommended final concentrations in the dishes (60.1 cm² for adherent
237 cells, 10 mL medium) are 50 µM and 200 µM *t*-BOOH for an incubation time of
238 30 min, 1 h and 2 h.
- 239 5. Harvest cells using trypsin as described in step 3. a – e.

- 240 a. Add 5 mL cold PBS + 5% FCS to the detached cells in suspension and transfer to
241 a 15 mL conical centrifuge tube
242 b. Centrifuge (5 min, 4 °C, 200 – 600 × g)
243 c. Remove the supernatant liquid
244 d. Resuspend the cell pellet in 1 mL cold PBS and transfer to a 1.5 mL reagent tube
245 yielding a pellet of approx. 4 – 8 × 10⁶ cells. The 15 mL tube should be rinsed with
246 additional ~ 300 µL cold PBS, which are transferred to same 1.5 mL reagent tube.
247 e. Centrifuge (5 min, 4 °C, 600 – 1000 × g) and remove the liquid
248 6. Freeze cell pellet at -80°C until analysis
249 7. Exclude cytotoxic effects of *t*-BOOH at the used concentrations and incubation times e.g.
250 by resazurin (Alamar Blue) assay (26) and lactate dehydrogenase assay (27, 28).

251 **QUESTION (1):** Why is it important to determine the cytotoxicity of the test compound at the
252 used concentration? Which parameters for different cytotoxic endpoints can be assessed?

253 **2.3 Cell culture assay: Inflammatory response in THP-1 cells**

- 254 1. The monocytic cell line THP-1 (obtained from the German Collection of Microorganisms
255 and Cell Cultures GmbH (DSMZ, Braunschweig, Germany)) is cultivated in suspension in
256 bicarbonate buffered RPMI 1640 medium supplemented with 10% FCS,
257 100 U/mL penicillin, 100 µg/mL streptomycin (P/S, 2%) and 2 mM L-glutamine (1%) in a
258 humidified incubator at 37 °C and 5% CO₂ in 10 cm dishes (58.8 cm²) for suspension cells
259 2. *Maintaining the cell culture:* Transfer cells every 4-5 days (Monday and Friday):
260 a. Gently resuspend cells in the dish and transfer cells and medium to a conical
261 centrifuge tube
262 b. Determine the cell number
263 c. Seed 1 × 10⁶ cells (on Monday and let them grow till Friday) or 1.8 × 10⁶ cells (on
264 Friday and let them grow till Monday) in a total of 10 mL fresh medium in new 10 cm
265 dishes (58.8 cm²) for suspension cells
266 3. Differentiation and incubation with test compounds:
267 a. For differentiation of cells, prepare medium by adding 10 µL of 50 µM vitamin D3
268 (in DMSO) and 10 µL of 1 µg/mL TGF-β1 (in PBS) to 10 mL RPMI 1640 medium
269 resulting in 50 nM vitamin D3 (0.1% DMSO) and 1 ng/mL TGF-β1. Seed cells at
270 densities of 0.125 × 10⁶ cells/mL in 10 mL of this medium in 10 cm dishes
271 (60.1 cm²) for adherent cells as cells will become partially adherent during
272 differentiation. Allow the cells to differentiate for 72 h

- 273 b. For incubation with test compounds, replace cell culture medium 7 h before the end
274 of the differentiation (65 h after the start of differentiation) with serum-free 50 mM
275 TRIS buffered RPMI medium (2% P/S, 1% L-glutamine):
- 276 i. Transfer non-attached cells and medium to a conical centrifuge tube and
277 add directly 5 mL serum-free 50 mM TRIS buffered RPMI medium (2% P/S,
278 1% L-glutamine) to the dish
 - 279 ii. Centrifuge the tube (5 min, room temperature, $500 \times g$) and remove the
280 medium
 - 281 iii. Add 5 mL serum-free 50 mM TRIS buffered RPMI medium (2% P/S, 1% L-
282 glutamine) to the cell pellet, resuspend cells and transfer back to the dish
 - 283 iv. Add 10 μL of the test compound (pharmacological drug/inhibitor) in DMSO
284 (e.g. add 10 μL 100 μM indomethacin yielding a final concentration of
285 100 nM) or DMSO (0.1%) as control. Suggestions for possible test
286 compounds are summarized in Tab. 1.
- 287 d. After 1 h of preincubation, add 10 μL 1 mg/mL LPS (in PBS) to the medium resulting
288 in 1 $\mu\text{g/mL}$ LPS (for control add 10 μL PBS) for 6 h

289 **QUESTION (2):** Which control incubations should be carried out to allow meaningful
290 interpretation? Which aspects should be considered when doing control incubations?

- 291 4. Harvest all adherent and non-adherent cells:
- 292 a. Scrape the cells from the dish in the culture medium with a cell scraper and
293 thoroughly flush the dish with the suspension
 - 294 b. Transfer cells and medium to a 15 mL conical centrifuge tube and place in ice
 - 295 c. Centrifuge (5 min, 4 °C, 200 – 600 $\times g$)
 - 296 d. Remove the medium
 - 297 e. Add 5 mL PBS and resuspend thoroughly
 - 298 f. Centrifuge and remove supernatant again
 - 299 g. Resuspend cell pellet in 1 mL PBS (containing 1% protease inhibitor mixture with
300 AEBSF, 129 Aprotinin, Bestatin, E-64, Leupeptin and Pepstatin A, if proteomics
301 analysis with the cell pellet is intended to be carried out)
 - 302 h. Transfer to a 1.5 mL reagent tube yielding a pellet of approx. $6 - 9 \times 10^6$ cells
 - 303 i. Centrifuge (5 min, 4 °C, 500 $\times g$) and remove the supernatant
 - 304 j. Freeze the cell pellet at -80 °C until analysis

305 5. Exclude cytotoxic effects of the test compounds at the used concentrations by resazurin
306 (Alamar Blue) assay (26) and lactate dehydrogenase assay (27, 28).

307 **2.4 Sample preparation**

308 *NOTE: Work on ice and store samples on ice during sample preparation to minimize artificial*
309 *oxylipin formation.*

310 *Free and total oxylipins, the protein concentration as well as total fatty acyls can be determined*
311 *from a single cell pellet.*

312 1. Resuspend cell pellets in an exact volume (300-500 μ L) PBS (containing 1% protease
313 inhibitor mixture with AEBSF, 129 Aprotinin, Bestatin, E-64, Leupeptin and Pepstatin A, if
314 proteomics analysis with the cell pellet is intended to be carried out) and add 10 μ L
315 inhibitor/antioxidant solution (0.2 mg/mL BHT, 100 μ M of the COX inhibitor indomethacin,
316 100 μ M of the soluble epoxide hydrolase inhibitor *trans*-4-[4-(3-Adamantan-1-yl-ureido)-
317 cyclohexyloxy]-benzoic acid (*t*-AUCB) and 100 μ M of the 15-LOX inhibitor BLX3887 in
318 methanol (MeOH))

319 2. Sonicate (keep the tube in ice during the sonication)

320 a. small ultrasonic tip (1 mm): amplitude 100%, unpulsed, cycle 1, 8 – 10 s

321 b. ultrasonic tip (3 mm): level 2, 20% output, unpulsed, 8 – 10 s

322 Repeat sonication for another 10 s if cell suspension doesn't look clear when viewed
323 towards the light

324 3. Determine protein content in an aliquot of the sonicate (use an exact volume of 10 – 30 μ L
325 for an appropriate dilution) *via* bicinchoninic acid (BCA) assay (29) allowing normalization
326 of the oxylipin concentration to the protein content. (Expected protein concentration in the
327 sonicate: for HCT-116 cells 3 – 6 mg/mL, for THP-1 cells 0.4 – 2 mg/mL). If the cell number
328 of the pellet is known the oxylipin concentration can alternatively be normalized on the cell
329 number.

330 For an assay as carried out in our lab with a working range of 0.1 – 1 mg/mL BSA the
331 following dilutions are suggested for the BCA assay if the cell pellet is resuspended in
332 500 μ L PBS (with inhibitors) for sonication:

333 a. For HCT-116 cells an 1+4 dilution

334 b. For THP-1 cells an 1+2 dilution

335 4. The homogenous cell lysate can be split to analyze free and total (the sum of free and
336 esterified) oxylipins from the same cell pellet depending on the scientific question. Based

337 on that the volume of this step should be decided, e.g. if higher sensitivity is needed for
338 free or total oxylipins.

339 If both free and total oxylipins are analyzed use 350 μL of the sonicate for the determination
340 of free oxylipins and 100 μL of the sonicate for the determination of total oxylipins and the
341 rest for protein determination *via* BCA. Otherwise select volumes to achieve best sensitivity
342 for the scientific question.

343 **QUESTION (3):** With the described methodology it is possible to analyze free and total
344 (the sum of free and esterified) oxylipins. Can you give examples in which experimental
345 setups the analysis of free or total oxylipins makes more sense and why? For the two
346 experiments described here is the analysis of free or total oxylipins more meaningful to
347 characterize the resulting biological effect?

348 a. For free oxylipins: Add the 2.8-fold volume of ice-cold methanol and 10 μL internal
349 standards (IS) (100 nM of $^2\text{H}_4$ -PGE₂, $^2\text{H}_4$ -TxB₂, $^2\text{H}_4$ -LTB₄, $^2\text{H}_8$ -5-HETE, $^2\text{H}_8$ -15-
350 HETE, $^2\text{H}_{11}$ -8(9)-EpETrE, $^2\text{H}_{11}$ -14(15)-EpETrE) using a repeating syringe dispenser
351 (“Hamilton-Repeater”) (500 μL syringe)

352 b. For total oxylipins: Add 400 μL ice-cold isopropanol and 10 μL internal standards
353 (IS) ($^2\text{H}_4$ -15-F_{2t}-IsoP, $^2\text{H}_{11}$ -5-(R,S)-5-F_{2t}-IsoP, $^2\text{H}_8$ -5-HETE, $^2\text{H}_8$ -15-HETE, $^2\text{H}_{11}$ -
354 8(9)-EpETrE, $^2\text{H}_{11}$ -14(15)-EpETrE) using a Hamilton-Repeater (500 μL syringe)

355 5. Vortex samples thoroughly and precipitate proteins by freezing at -80 °C for at least 30 min

356 *Possible Break for hours up to several days*

357 6. When samples are taken from -80 °C freezer, leave them at room temperature for 1-2 min,
358 then vortex briefly

359 7. Centrifuge (10 min, 4 °C, 20 000 x g)

360 *OPTION: The pellet is discarded or can be used to determine protein levels using targeted*
361 *LC-MS/MS based proteomics analysis as described in (14). Therefore, we recommend to*
362 *centrifuge with a slightly lower acceleration and to store the pellet at -80 °C until proteomics*
363 *analysis.*

364 8. The supernatant serves as sample for oxylipin analysis:

365 a. For free oxylipins: The sample can be loaded directly onto the prepared SPE
366 cartridge (see Section 2.5 “Solid Phase Extraction”, step 3)

367 b. For total oxylipins a hydrolysis step is required

368 9. Hydrolysis for total oxylipins:

369 a. Transfer supernatant into a 1.5 mL reagent tube

370 b. Add 100 μL 0.6 M KOH in MeOH/H₂O (75/25, v/v)

- 371 c. Vortex
- 372 d. Hydrolyze sample for 30 min at 60 °C using a pre-heated shaker (500 rpm)
- 373 e. After hydrolysis, cool sample immediately on ice and neutralize by adding 20 µL
- 374 acetic acid (HAc, 25% in water)
- 375 f. Vortex
- 376 g. Centrifuge sample very briefly in order to collect the liquid on the bottom of the
- 377 reaction tube
- 378 *OPTION:* Fatty acyls can be determined in an aliquot of the sample according to (30). For free
- 379 fatty acyls use an appropriate aliquot of the sample before hydrolysis (e.g. dilute 50
- 380 µL + 50 µL EtOH). For total fatty acyls use an appropriate aliquot of the hydrolyzed
- 381 sample (e.g. dilute 20 µL + 180 µL EtOH for low abundant fatty acyls and 10 µL +
- 382 490 µL EtOH for high abundant fatty acyls).

383 **2.5 Solid Phase Extraction (SPE)**

- 384 1. Prepare SPE cartridges (anion exchange Bond Elut Certify II SPE cartridges 200 mg,
- 385 3 mL, Agilent, Waldbronn, Germany) by washing with
- 386 a. One column volume of ethyl acetate/*n*-hexane (75/25, v/v) containing 1% HAc
- 387 b. One column volume of MeOH
- 388 c. One column volume 0.1 M disodium hydrogen phosphate (Na₂HPO₄) buffer in
- 389 water/MeOH (95/5, v/v) (adjusted to pH 6.0 with HAc). Close the valve when the
- 390 solution is 2-3 mm above the stationary phase
- 391 2. Add 2.0 mL 0.1 M Na₂HPO₄ buffer in water (adjusted to pH 6.0 with HAc) to the cartridges
- 392 3. Load samples with a pasteur pipette and mix thoroughly with the buffer
- 393 *NOTE: The content of organic solvent on the cartridge should be kept below 16% to*
- 394 *prevent breakthrough/elution of the analytes during the loading step, if necessary*
- 395 *the volume of the buffer needs to be adjusted or the sample needs to be evaporated*
- 396 *appropriately.*
- 397 4. Check pH using pH stripes (5.1-7.2 scale); only if necessary, carefully adjust pH to 6.0 with
- 398 diluted HAc (if HAc has to be added, only few µl are needed)
- 399 5. Open valves and let samples run by gravity until completely sunk into the stationary
- 400 phase
- 401 6. Wash with
- 402 a. One column volume water
- 403 b. One column volume MeOH/H₂O (50/50, v/v)
- 404 7. Dry samples with vacuum:

- 405 a. Close valves of all cartridges and create stable -200 mbar negative pressure
406 within the manifold
- 407 b. Open valves of two or three samples for drying the cartridges
408 *NOTE: Drying of the cartridges can be verified by putting a pasteur pipette cone*
409 *on top of the cartridges (cone should tighten)*
- 410 c. Close valves after 30 s (it is not critical if samples dry a few seconds longer,
411 however, do not dry them longer than 1 min)
- 412 d. Repeat drying step for all samples in pairs of three or two
- 413 8. Elute analytes with 2.0 mL of 75/25 (v/v) ethyl acetate/*n*-hexane with 1% HAc by gravity in
414 glass tubes containing 6 µL 30% glycerol in MeOH (a dispenser resistant to organic
415 solvents can be used to measure the eluent volume). Remove last drops of eluent from
416 stationary phase by applying positive pressure with the pasteur pipette cone at the top of
417 the cartridge
- 418 9. Evaporate samples to dryness using a vacuum centrifuge (1 mbar, 30 °C, ~60 min)
- 419 10. Reconstitute samples in 50 µL MeOH using a Hamilton-Repeater (use 2.5 mL syringe)
420 and dissolve samples by sonication and vortexing
- 421 *NOTE: Inclusion of one or more secondary internal standards in the reconstitution*
422 *solvent enables to calculate the extraction efficiency of the sample preparation*
423 *(see section 3.1) (for examples of secondary internal standards see (25)).*

424 **QUESTION (4):** How would you select a secondary internal standard?

- 425 11. Transfer samples completely into 1.5 mL reagent sample tubes
- 426 12. Freeze samples at -80 °C for at least 30 min
- 427 *Possible break: Reconstituted samples can be stored for at least 2 months at -80 °C with only*
428 *slight changes in the oxylipin pattern (< 20% for most analytes, CAVE: quantification of*
429 *isoprostanes might be impaired)*
- 430 13. Centrifuge (10 min, 4 °C, 20 000 x g)
- 431 14. Transfer clear (!) supernatant into vial with insert. Centrifuge samples again if
432 supernatant is not completely clear

433 **2.6 LC-ESI(-)-MS/MS analysis**

434 The analysis is carried out using reversed-phase (RP) LC-MS/MS on a triple quadrupole
435 instrument operated in selected reaction monitoring mode (SRM frequently termed MRM)
436 following negative electrospray ionization (ESI).

437 The selection of transitions, as well as electronical parameters and source settings requires
438 extensive optimization, which is described in detail in (23, 24).

439 An excellent chromatographic separation is also mandatory for oxylipin analysis because several
440 oxylipins cannot be separated by MS/MS (a list of typical critical separation pairs can be found in
441 the supplementary information of (31).

442 Here, we describe an optimized method using a liquid chromatography system composed of a
443 1290 Infinity LC system (Agilent, Waldbronn, Germany) with autosampler, binary pump and
444 column oven coupled to a QTRAP mass-spectrometer (Sciex, Darmstadt, Germany). For data
445 acquisition and instrument control Analyst Software and for integration and quantification
446 Multiquant Software is used.

447 However, when using (slightly) different instrumentation, the LC-MS/MS parameters need to be
448 adapted to the used system.

449

- 450 1. Inject (5 μ L) the samples into the LC-MS/MS system (keep samples in a 4 °C cooled
451 autosampler until injection)
- 452 2. For liquid chromatography an UPLC system is needed as a backpressure of about 600 bar
453 is reached during the analysis. Separate analytes on a Zorbax Eclipse Plus C-18 reversed-
454 phase column (2.1 \times 150 mm, particle size 1.8 μ m; RRHD; Agilent, Waldbronn, Germany)
455 equipped upstream with an inline filter (0.3 μ m, 1290 infinity II in-line filter, Agilent,
456 Waldbronn, Germany) and a SecurityGuard Ultra C18 cartridge as pre-column (2.1 \times
457 2 mm, Phenomenex, Aschaffenburg, Germany) at 40 °C and a flow rate of 0.3 mL/min
- 458 3. For chromatographic separation use the following binary gradient with solvent A 0.1%
459 HAc/solvent B (95/5, v/v), and solvent B acetonitrile/MeOH/HAc (800/150/1, v/v/v):
460 0-1.0 min isocratic 21% B, 1.0-1.5 min linear from 21% B to 26% B, 1.5-10.0 min linear
461 from 26% B to 51% B, 10.0-19 min linear from 51% B to 66% B, 19-25.1 min linear from
462 66% B to 98% B, 25.1-27.6 min isocratic 98% B, 27.6-27.7 min linear from 98 B to 21% B
463 followed by reconditioning for 3.8 min. During the first 2 min and the last 6 min of each run,
464 the LC flow is directed to waste using the 2-position-6-port valve integrated in the MS
- 465 4. For mass spectrometric detection use negative electrospray ionization (ESI-) with the
466 following source settings: ion-spray voltage: -4500 V, curtain gas (N₂): 35 psi, nebulizer
467 gas (gas 1, zero air): 60 psi generated with a zero air generator, and drying gas (gas 2,
468 zero air): 60 psi at a temperature of 475 °C. The offset of the sprayer is 0.250 cm for the
469 vertical axis and 0.550 cm for the horizontal axis, the electrode protrusion is approx. 1 mm
- 470 5. For detection of the analytes use scheduled selected reaction monitoring mode (SRM) with
471 nitrogen as collision gas (set to "high", 12 psi) and a detection window of \pm 22.5 s around
472 the expected retention time and a cycle time of 0.4 s

- 473 6. For each analyte use the optimized compound-specific parameters. Optimized parameters
474 for the selected analytes evaluated within the experiments are summarized in Tab 2. A list
475 of all covered analytes and their mass spectrometric parameters can be found in (14)
476 7. For calibration, mix stock solutions of the individual authentic standards (in MeOH) and
477 dilute in glass volumetric flasks (5 – 100 mL) with MeOH at 10 concentration levels (0.1,
478 0.25, 0.5, 1, 2, 5, 10, 20, 100 and 500 nM), each with 20 nM of the internal standards. An
479 appropriate procedure is described in detail in (14, 25)
480 *NOTE: For evaluating the extraction efficiency of the sample preparation based on*
481 *secondary internal standards a secondary calibration curve should be prepared*
482 *(covering 20 – 120% recovery of IS1).*
483 8. Analyze calibration standards with the same method as samples
484

485 **Step-by-step protocol ends**
486

487 **3 Data analysis/interpretation**

488 In contrast to non-targeted metabolomics, with targeted metabolomics the metabolites in the
489 biological samples are quantified. Thus, the key step is the calculation of concentrations of the
490 analyzed compounds. Similar to all other chromatography-mass spectrometry-based quantitative
491 methods used in food, pharmaceutical, forensic or environmental chemistry and other fields the
492 concentration is calculated based on external calibration. In order to compensate for losses
493 occurring by sample preparation, matrix effects and instability of the MS-signal internal standards
494 (IS) are used. Thus, instead of the peak area of the analytes, the peak area ratio of the analyte to
495 its assigned internal standard is used.

496 **QUESTION (5):** What is ion suppression occurring in ESI-MS and how do you detect it? Why are
497 isotopically labeled standards the only way of allowing a robust quantification of analytes in
498 biological samples with varying matrix (matrix = all other compounds in the sample)?

499 **3.1 Quantification by external calibration using internal standards**

- 500 1. Integrate peak areas of analytes and internal standards
501 2. Determine areas of the analytes and corresponding internal standards (Tab. 2) and
502 calculate area ratios

- 503 3. Determine the limit of detection (LOD; $S/N \geq 3$) and the lower limit of quantification (LLOQ;
504 $S/N \geq 5$ and accuracy of the calibration level 80-120%, see step 4 c) based on the signal-
505 to-noise ratio of the peak
- 506 4. Determine the calibration curve
- 507 a. Plot the peak area ratio at the individual calibration levels against the respective
508 concentration and determine the calibration curve using linear least square
509 regression (weighting: $1/x$ or $1/x^2$)
- 510 b. Verify the absence of signal saturation: If the signal of the highest calibration level
511 is below the linear calibration curve, this concentration is not within the linear range,
512 but above the upper limit of quantification (ULOQ) and should be removed from the
513 calibration. Otherwise the highest injected calibration level limits the upper linear
514 range, do not extrapolate
- 515 c. Use the calibration curve to calculate the concentration of the calibration levels.
516 Evaluate the quality of the calibration curve by determination of the accuracy of the
517 determined vs. the theoretical concentration at each calibration level. For all
518 calibration levels the accuracy should be within $100 \pm 15\%$, for the LLOQ
519 $100 \pm 20\%$. Otherwise repeat the calibration
- 520 *NOTE: Quantification can also be performed without weighting, however using the suggested*
521 *$1/x$ or $1/x^2$ weighting more accurate results will be obtained at low concentrations*
- 522 *OPTION: For evaluation of sample preparation – more specifically of the extraction efficiency*
523 *– determine a second calibration curve using the area ratio of internal standards*
524 *added at the beginning of sample preparation to assigned secondary internal*
525 *standards (IS2) used for sample reconstitution. Only samples with a sufficient*
526 *predefined extraction efficiency should be further evaluated.*
- 527 5. For quantification, calculate the oxylipin concentration in the vial based on the analyte to
528 corresponding IS area ratio in the samples using the external calibration of the respective
529 authentic standards (see step 4). Only determine concentrations between the LLOQ and
530 the highest calibration point/ULOQ to ensure reliable quantification (as indicated in (32)).
531 Suggested unit: nmol/L (= nM) or pmol/mL
- 532 6. Taking the determined protein concentration, the volumes used for homogenization and
533 oxylipin analysis and the reconstitution volume after SPE into account calculate the
534 concentration of oxylipins in the cells, e.g. in pmol/mg protein (14). Alternatively, the
535 oxylipin concentration can be calculated based on the number of cells, eg in fmol/ 10^6 cells
536 (12)

- 537 7. For epoxy-fatty acid regioisomers, two peaks can be detected with *trans*-epoxy-PUFA
538 isomers eluting 0.14 – 0.3 min after their corresponding *cis*-isomers
539 a. Characterize *trans*-epoxy-PUFA based on retention time and identical MS-
540 fragmentation pattern as described (12, 33-35).
541 b. For the individual regioisomers determine the *trans/cis*-epoxy-PUFA ratio
542

543 3.2 Evaluation of the quantitative data

- 544 1. Keep in mind what is (your/the) hypothesis of the experiment: Which oxylipins do you
545 expect to be up/down regulated by the stimuli used? Which experimental groups are you
546 going to select as controls to compare the impact of the stimuli?

547 *NOTE: Further data evaluation and statistical analyses can be performed using e.g. GraphPad*
548 *Prism, R, origin, SPSS or a similar software. Data evaluation can also be carried out*
549 *using Microsoft Excel, however, it has limited functions for statistical analysis.*

- 550 2. For each set of samples (the replicates of the same treatment) calculate the mean and the
551 standard deviation (SD) and relative standard deviation (RSD).

552 *NOTE: For specific pairs of oxylipins where a ratio is biologically meaningful to be*
553 *calculated (e.g. the ratio between cis- and trans epoxy PUFA), calculate the ratio*
554 *for each sample. Then the mean and SD is calculated leading to a higher precision.*

- 555 • Plot the results in diagrams allowing to compare the concentration of individual oxylipins
556 or ratio of oxylipin pairs with the controls (mean \pm SD) (examples can be found in (12, 14)).
 - 557 ○ HCT-116/*t*-BOOH: Compare the different *t*-BOOH concentrations against the
558 control without *t*-BOOH. Which oxylipins are elevated, which are not changed? Is
559 it possible to deduce a dose- and time-dependent effect of *t*-BOOH incubation?
560 Correlate different oxylipins which may result from non-enzymatic conversion
561 against each other.
 - 562 ○ THP-1/LPS: Which oxylipins are not affected and which are elevated in incubations
563 with LPS (positive control) vs. without LPS (negative control)? Can the test
564 compounds block the LPS-induced formation of oxylipins? Are there concentration-
565 dependent effects of the test compounds?

566 3.3 Interpretation and learnings

567 **Oxidative stress:** Oxidative stress leads to the formation of reactive oxygen species. Among
568 other cellular biomolecules, lipids are oxidized by radical chain reactions.

569 Both, living cells as well as in foods (e.g. plant oils) particularly unsaturated fatty acids are prone
570 to oxidation, a process called aut(o)oxidation. Here, initially hydroperoxy radicals are formed which
571 further react to volatile aldehydes such as MDA but also to the oxylipins measured in this
572 experiment. The formation of hydroxy-PUFA from hydroperoxy-PUFA occurs by reduction, e.g.
573 catalyzed by cellular glutathione peroxidases. The level of hydroxy-PUFA thus should be analyzed
574 as potential marker of oxidative stress. Besides hydroxy-PUFA also other oxylipins are elevated
575 by oxidative stress. With the targeted oxylipin metabolomics approach several different oxylipin
576 classes are analyzed in parallel.

577 In this experiment, different oxylipin classes should be evaluated with respect to their changes
578 during *t*-BOOH-induced oxidative stress. Based on the literature (10-12, 23), understand how
579 prostanoid-like isoprostanes are formed during oxidative stress, and suggest potential
580 mechanisms how *trans*-epoxy-PUFA can be formed. Evaluate which oxylipin class, and which
581 isomers within this class are strongest elevated by *t*-BOOH-induced oxidative stress. Suggest
582 which (set of) oxylipins could be used to assess autoxidation in diseases associated with oxidative
583 stress and how these experiments/sampling could be carried out.

584
585 The first set of experiments demonstrates that PUFA are non-enzymatically oxidized to oxylipins.
586 However, oxylipins are also formed by specific enzymes giving rise to highly potent lipid mediators.
587 In the experiments with THP-1/LPS the focus is set on the 5-LOX- and COX-derived oxylipins,
588 which are formed in macrophage-like cells upon an inflammatory stimulus.

589
590 **Question (6):** What is LPS and how does it elicit an inflammatory stimulus in mammalian cells?

591
592 The 5-LOX- and COX-derived oxylipins and their modulation by LPS treatment should be
593 evaluated. Is there an overlap to oxylipins formed by autoxidation?

594
595 **5-LOX:** Which oxylipins can be used to monitor 5-LOX activity? What is the difference between
596 the enzymatic and the non-enzymatic LTA₄ products (Fig. 1)? How can the level of the different
597 LTB₄ isomers be interpreted with respect to the LTAH activity in the cells?

598
599 **COX:** Inflammation is a major target of today's pharmaceuticals. Among the pharmaceuticals sold
600 over-the-counter, NSAIDs (ibuprofen, indomethacin, diclofenac) are under the top sellers in
601 pharmacies in Europe and worldwide. These compounds directly target COX enzymes and elicit
602 their intended effects by reducing the formation of pain-mediating and fever-causing prostanoids.

603 However, these compounds block both COX-1 and COX-2 activity, causing severe side effects,
604 such as ulcers in the gastrointestinal tract by COX-1 inhibition. In order to selectively block COX-2,
605 selective inhibitors have been developed such as celecoxib. The strongest agent to dampen
606 inflammation are steroids, such as the synthetic glucocorticoid dexamethasone, which is e.g. used
607 in severe Covid-19.

608 In the experiments these two classes of drugs are tested for their effects on the inflammatory
609 response. What are the similarities, what are the differences in the effects of the compounds
610 (classes) on the oxylipin formation particularly based on COX activity in the employed model of
611 human macrophages? Also, compare the concentration range in which the compounds are active.
612 If it is possible to analyze a sufficient number of different concentrations, calculate a dose-
613 response curve and compare the IC₅₀ values. Based on the potency and selectivity on 5-LOX,
614 COX(s) and autoxidation, this assay also allows to evaluate the inhibitory activity of new
615 compounds.

616

617 **Take home message starts**

618 **4 Take home message**

- 619 • Oxylipins are formed non-enzymatically by (lipid) autoxidation as well as by specific
620 enzymatic reactions. In mammals, enzymatic formation comprises three main pathways
621 catalyzed by cyclooxygenases (COX), lipoxygenases (LOX) and cytochrome P450
622 monooxygenases. The initial products are further converted by several other enzymes
623 forming specific lipid mediators with distinct biological activity.
- 624 • Multiple oxylipins can be formed by several pathways. For example, 15-HETE can be
625 formed by LOX and as side product by COX, and also as minor product by CYP.
626 Moreover, 15-HETE is a major product of autoxidation.
- 627 • Autoxidation leads to a multitude of different oxylipins: Hydroxy fatty acids such as 15-
628 HETE, *trans*-epoxy-fatty acids as well as isoprostanes serve as marker of oxidative
629 stress. Thus, the parallel analysis of different oxylipins can serve together as markers of
630 oxidative stress (see (12) for further information).
- 631 • Several oxylipins are highly potent lipid mediators, playing a key role in the regulation of
632 pain, fever and inflammation. Thus, oxylipin formation is a major drug target, e.g. for
633 COX-inhibiting non-steroidal anti-inflammatory drugs (NSAID) as shown in the
634 inflammatory assay.

635 • Oxylipins occur in biological samples in low concentrations, in cells a range of pmol to
636 nmol per mg protein can be expected. Targeted liquid chromatography-tandem mass
637 spectrometry (LC-MS/MS) is currently the best-suited method to quantify the low
638 abundant lipid mediators. Quantification is carried out by external calibration with internal
639 standards. In the experiments it becomes clear, that basic method validation is
640 indispensable including the definition of upper and lower limits of quantification, accuracy
641 and recovery rates.

642 **Take home message ends.**

643

644 **Tables and Figures**645 **Table 1: Suggested test compounds and concentrations for the incubation to assess inflammatory response in THP-1 cells.**

Test compounds	Concentration range
Dexamethasone	10 pM – 1 μM
Indomethacin	1 nM – 10 μM
Celecoxib	1 nM – 10 μM

646
647 **Table 2: Parameters for the targeted oxylipin metabolomics LC-ESI(-)-MS/MS analysis of selected oxylipins covered in the**
648 **experiments.** Shown are the mass transitions for quantification in scheduled SRM mode (*m/z* of precursor and fragment ion). The
649 instrument settings (declustering potential (DP), entrance potential (EP), collision energy (CE), collision cell exit potential (CXP)) are
650 shown for a Sciex 5500 QTrap instrument. Moreover, the internal standard (IS) for each analyte, as well as the retention time (*t*_R), limit
651 of detection (LOD), the calibration range (lower limit of quantification (LLOQ), upper limit of quantification (ULOQ)) of our method are
652 provided as orientation for method development.

analyte	mass transition		mass parameters				internal standard	<i>t</i> _R [min]	calibration range		
	Q1	Q3	DP [V]	EP [V]	CE [V]	CXP [V]			LOD [nM]	LLOQ [nM]	ULOQ [nM]
15-F _{2t} -IsoP (8- <i>iso</i> -PGF _{2α})	353.1	193.1	-95	-10	-34	-8	² H ₄ -8- <i>iso</i> -PGF _{2α}	7.58	0.10	0.25	500
TxB ₂	369.2	169.1	-80	-10	-24	-7	² H ₄ -TxB ₂	7.68	0.25	0.50	1000
5(<i>R,S</i>)-5-F _{2t} -IsoP (5- <i>i</i> PF _{2α} -VI)	353.2	114.8	-85	-10	-27	-8	² H ₄ -8- <i>iso</i> -PGF _{2α}	8.07	0.1	0.25	500
PGE ₂	351.2	271.3	-80	-10	-23	-6	² H ₄ -PGE ₂	8.91	0.25	0.50	750
6- <i>trans</i> -LTB ₄	335.2	195.1	-80	-10	-22	-9	² H ₄ -LTB ₄	13.36	0.1	0.25	250
6- <i>trans</i> -12- <i>epi</i> -LTB ₄	335.2	195.1	-85	-10	-20	-9	² H ₄ -LTB ₄	13.51	0.1	0.25	500
LTB ₄	335.2	195.1	-80	-10	-22	-9	² H ₄ -LTB ₄	13.83	0.05	0.1	500

5(S),12(S)-DiHETE	335.2	195.1	-80	-10	-21	-8	² H ₄ -LTB ₄	14.40	0.025	0.05	500
12-HHTrE	279.1	179.0	-70	-10	-15	-8	² H ₁₁ -11,12-DiHETrE	15.62	0.25	0.5	500
5(S),6(R)-DiHETE (ARA)	335.2	115.1	-70	-10	-20	-8	² H ₁₁ -11,12-DiHETrE	17.33	0.020	0.039	390
5(S),6(S)-DiHETE (ARA)	335.2	115.1	-70	-10	-20	-8	² H ₁₁ -11,12-DiHETrE	17.80	0.022	0.045	223
15-HETE	319.2	219.2	-80	-10	-18	-8	² H ₈ -15-HETE	20.08	0.11	0.22	220
11-HETE	319.2	167.2	-80	-10	-21	-7	² H ₈ -12-HETE	20.68	0.022	0.044	219
9-HETE	319.2	167.2	-80	-10	-21	-7	² H ₈ -5-HETE	21.45	0.27	0.4	265
5-HETE	319.2	115.2	-80	-10	-19	-7	² H ₈ -5-HETE	21.74	0.018	0.035	350
14(15)-EpETrE	319.2	219.2	-90	-10	-15	-4	² H ₁₁ -14(15)-EpETrE	22.45	0.1	0.25	500
<i>trans</i> -14(15)-EpETrE	319.2	219.2	-90	-10	-15	-4	² H ₁₁ -14(15)-EpETrE	22.60		relative quantification based on 14(15)-EpETrE	
11(12)-EpETrE	319.2	167.2	-85	-10	-16	-7	² H ₁₁ -8(9)-EpETrE	22.98	0.05	0.1	500
<i>trans</i> -11(12)-EpETrE	319.2	167.2	-85	-10	-16	-7	² H ₁₁ -8(9)-EpETrE	23.13		relative quantification based on 11(12)-EpETrE	
8(9)-EpETrE	319.2	155.2	-90	-10	-16	-6	² H ₁₁ -8(9)-EpETrE	23.16	0.25	0.5	500
<i>trans</i> -8(9)-EpETrE	319.2	155.2	-90	-10	-16	-6	² H ₁₁ -8(9)-EpETrE	23.31		relative quantification based on 8(9)-EpETrE	
² H ₄ -8- <i>iso</i> -PGF _{2α}	357.2	196.8	-75	-10	-34	-8	<i>internal standard</i>	7.55			
² H ₄ -TxB ₂	373.3	173.2	-85	-10	-23	-8	<i>internal standard</i>	7.66			
² H ₄ -PGE ₂	355.2	275.3	-80	-10	-24	-6	<i>internal standard</i>	8.88			
² H ₁₁ -8,12- <i>iso</i> -iPF _{2α} -VI	364.2	320.1	-90	-10	-30	-8	<i>internal standard</i>	10.01			
² H ₄ -LTB ₄	339.2	197.2	-80	-10	-22	-9	<i>internal standard</i>	13.76			
² H ₈ -15-HETE	327.2	226.0	-90	-10	-18	-8	<i>internal standard</i>	19.88			
² H ₈ -12-HETE	327.2	184.2	-85	-10	-20	-8	<i>internal standard</i>	20.93			

² H ₈ -5-HETE	327.2	116.1	-80	-10	-19	-8	<i>internal standard</i>	21.60
² H ₁₁ -14(15)-EpETrE	330.2	219.3	-90	-10	-16	-4	<i>internal standard</i>	22.32
² H ₁₁ -8(9)-EpETrE	330.2	155.0	-80	-10	-16	-7	<i>internal standard</i>	23.05
Aleuritic Acid	303.1	268.8	-75	-10	-41	-10	<i>secondary internal standard</i>	5.39
(1-(1-(Ethyl-sulfonyl)piperidin-4-yl)-3-(4-(trifluoromethoxy)phenyl)urea)	394.0	176.0	-100	-10	-21	-10	<i>secondary internal standard</i>	10.69
12-oxo Phytodienoic acid (OPDA)	291.1	165.0	-90	-10	-27	-10	<i>secondary internal standard</i>	15.56
12-[[[tricyclo[3.3.1.1 ^{3,7}]dec-1-ylamino)carbonyl]amino]-dodecanoic acid (AUDA)	391.0	240.1	-100	-10	-25	-10	<i>secondary internal standard</i>	19.66

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654

655 **Figure Captions**

656 **Fig. 1: Simplified overview of oxylipin formation.** Shown are the main pathways of
657 arachidonic acid-derived eicosanoid formation by COX, LOX and CYP as well as non-enzymatic
658 autoxidative formation. Formation of enzymatic products is illustrated by green arrows,
659 autoxidation by orange arrows, black arrows depict chemical non-enzymatic breakdown.
660 Furthermore, modulation of eicosanoid products by the differentiation and stimulation used in the
661 experiment to investigate inflammatory response (THP-1 assay) is indicated.

662 **Fig. 2: Workflow of cell incubation and sample preparation.** Shown is the incubation strategy
663 used in the experiments in this chapter: Left: Induction of oxidative stress by *t*-BOOH in adherent
664 HCT-116 cells; Right: Differentiation and compound testing in the context of inflammation
665 induced by LPS using the monocytic cells THP-1. Bottom: Sample preparation strategy for
666 analysis of free and esterified oxylipins. The possible parallel analysis of protein levels as well as
667 fatty acyls is highlighted.

668

669 **Literature**

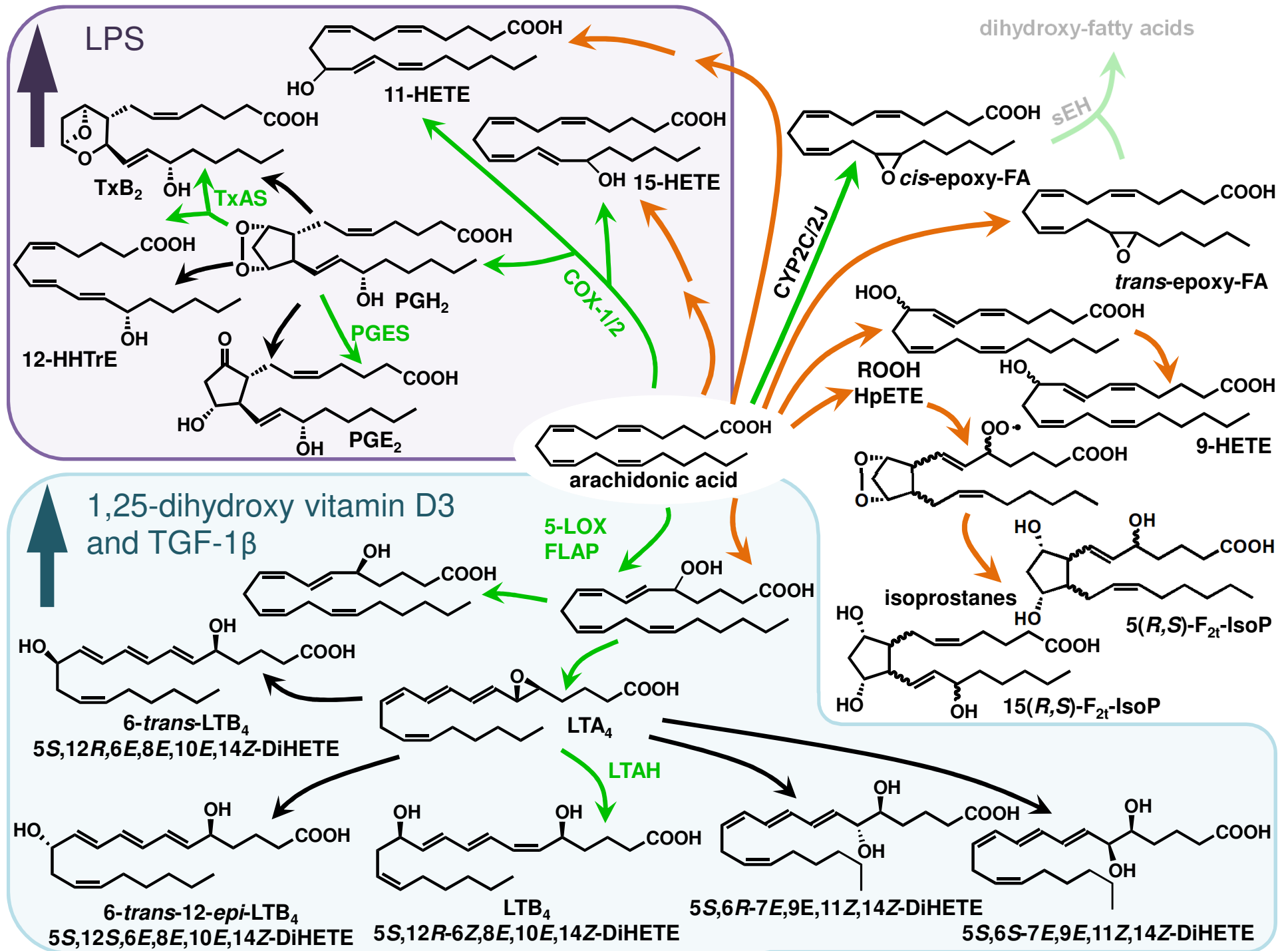
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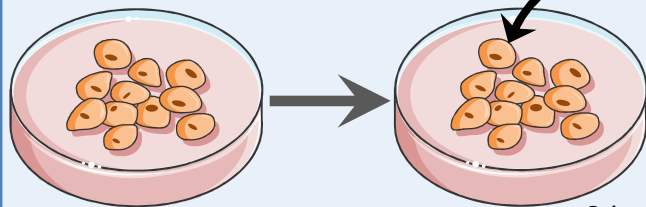


Cell culture: oxidative stress

HCT-116

induction of oxidative stress

tert-butyl hydroperoxide
(50 μ M and 200 μ M)



2 h

24 h

1 h

0.5 h

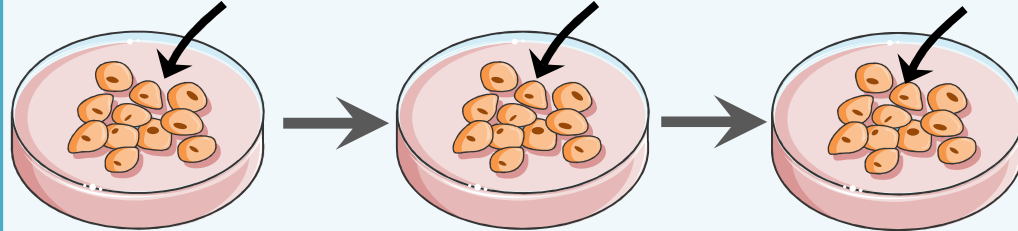
Cell culture: inflammatory response

Monocytic cells: THP-1

vitamin D3 (50 nM 0.1% DMSO),
1 ng/mL TGF- β 1 for 65 h

test compound

1 μ g/mL LPS



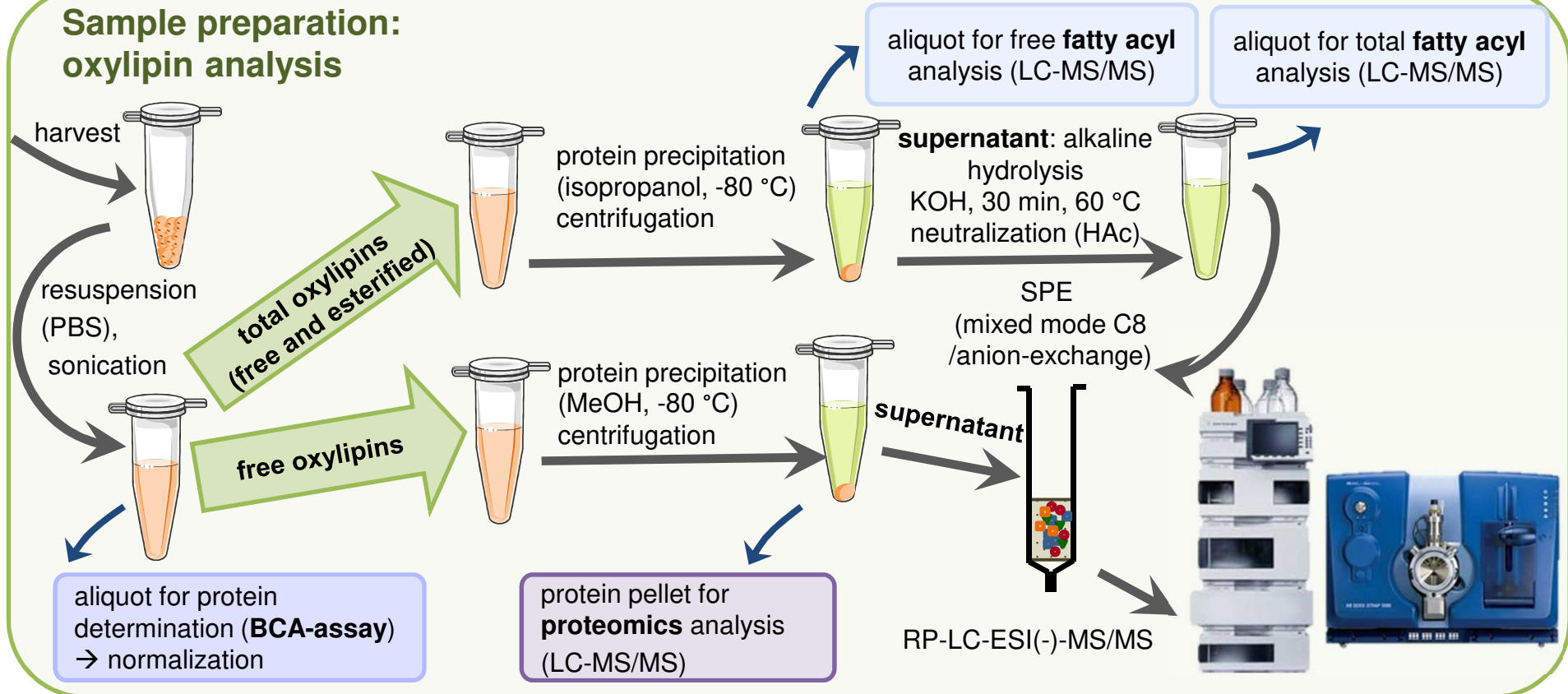
65 h

72 h

1 h

6 h

Sample preparation: oxylipin analysis



Supplementary Information

Quantitative analysis of eicosanoids and other oxylipins

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Investigation of oxidative stress and inflammation by means of targeted metabolomics of oxylipins in cell culture

Katharina M. Rund and Nils Helge Schebb

Chair of Food Chemistry, Faculty of Mathematics and Natural Sciences, University of Wuppertal, Gaußstr. 20, 42119 Wuppertal

Preparation of solutions

Solutions for cell culture experiments

- **DMEM supplemented with 10% fetal calf serum (FCS), 100 U/mL penicillin, 100 µg/mL streptomycin (P/S, 2%) and 2 mM L-glutamine** (for adherent HCT-116 cells)
For 1 L dissolve 3.7 g NaHCO₃ and 10 g DMEM (powder 1000 mg/L glucose and L-glutamine, without sodium bicarbonate) in 900 mL water (ultrapure 18 MΩ*cm), adjust pH to 6.8 and fill it up to 1 L. Sterile filtrate the solution. Add 100 mL FCS, 20 ml P/S and 10 mL 200 mM L-glutamine to 870 mL prepared DMEM.
- **RPMI 1640 medium supplemented with 10% fetal calf serum (FCS), 100 U/mL penicillin, 100 µg/mL streptomycin (P/S, 2%) and 2 mM L-glutamine (1%)** (for maintaining suspension cell line THP-1)
For 1 L add 100 mL FCS, 20 ml P/S and 10 mL 200 mM L-glutamine to 870 mL RPMI 1614 (without L-glutamine, with sodium bicarbonate, liquid).
- **Serum-free 50 mM TRIS buffered RPMI medium (2% P/S, 1% L-glutamine)**
For 1 L add 10.39 g RPMI 1614 (with L-glutamine, without sodium bicarbonate, powder) and 6.05 g TRIS to 0.9 L water (ultrapure 18 MΩ*cm) and adjust pH to 7.4 with HCl. Sterile filtrate the solution. Add 20 mL P/S to 980 mL prepared medium.
- **FCS with 10% DMSO**
Add 1 mL DMSO to 9 mL FCS.
- **Phosphate buffered saline (PBS)**
For 1 L add 5.84 g NaCl, 0.34 g KCl, 0.99 g Na₂HPO₄ (anhydrous), 0.41 g KH₂PO₄ (anhydrous) to 500 mL water (ultrapure 18 MΩ*cm) in a beaker. Dissolve the salts and adjust pH to 7.4 by addition of HCl or NaOH. Transfer the solution to a 1 L volumetric flask and fill it up to the mark with water (ultrapure 18 MΩ*cm). Sterile filtrate the solution.
- **PBS-EDTA (500 mM EDTA in PBS)**
For 1 L add 5.84 g NaCl, 0.34 g KCl, 0.99 g Na₂HPO₄ (anhydrous), 0.41 g KH₂PO₄ (anhydrous) and 0.2 g EDTA-disodium salt (dihydrate) to 500 ml water (ultrapure 18 MΩ*cm) in a beaker. Dissolve the salts and adjust pH to 7.4 by addition of HCl or NaOH. Transfer the solution to a 1 L volumetric flask and fill it up to the mark with water (ultrapure 18 MΩ*cm). Sterile filtrate the solution.

- **Trypsin (0.25% trypsin in PBS-EDTA)**
Add 100 mL 25 g/L trypsin to 900 mL PBS-EDTA.
- **PBS with 5% FCS**
Add 5 mL FCS to 95 mL PBS.
- **PBS + 1% protease inhibitor**
Add 100 µL protease inhibitor mixture to 10 mL PBS.
- **vitamin D3**
Stock: 50 mM in DMSO.
Dilution for experiments: 50 µM in DMSO. Add 10 µL 50 mM vitamin D3 to 90 µL DMSO.
Add 10 µL 5 mM vitamin D3 to 990 µL DMSO. Store aliquots à 200 µL.
- **TGF-β1**
Stock: 0.1 mg/mL in 10 mM citric acid.
Dilution for experiments: 1 µg/mL in 0.1% BSA in PBS. Add 20 µL 0.1 mg/mL TGF-β1 to 1980 µL 0.1% BSA in PBS. Store aliquots à 200 µL.
- **LPS**
Stock: 10 mg/mL in PBS.
Dilution for experiments: 1 mg/mL in PBS. Add 300 µL 10 mg/mL LPS to 2.7 mL PBS.
Store aliquots à 500 µL.
- **Test compounds**, e.g. indomethacin, dexamethasone, celecoxib
Prepare suitable concentration in DMSO

Solutions for SPE

- **Antioxidant-Mixture** (0.2 mg/mL BHT, 100 µM sEHi (*t*-AUCB), 100 µM indomethacin, 100 µM BLX3887 in MeOH):
 - a) BHT (0.2 mg/mL): Dissolve 10 mg BHT in 20 mL MeOH (LC-MS grade)
 - b) *t*-AUCB: Dissolve 4.13 mg *t*-AUCB in 1 mL DMSO for a 10 mM stock
 - c) Indomethacin: Dissolve 3.58 mg Indomethacin in 1 mL DMSO for a 10 mM stock
 - d) BLX3887: Dissolve 3.085 mg BLX3887 in 1 mL DMSO for a 10 mM stock
 Add inhibitors indomethacin, BLX3887 and *t*-AUCB to BHT solution (final inhibitor concentration 100 µM):
 - 1940 µL BHT (0.2 mg/mL)
 - + 20 µL 10 mM *t*-AUCB
 - + 20 µL 10 mM indomethacin
 - + 20 µL 10 mM BLX3887
- **Buffer for conditioning of SPE cartridges: 0.1 M Na₂HPO₄/HAc (pH 6.0) in H₂O/MeOH (95/5, v/v)**
Dissolve 1.78 g disodium phosphate dihydrate in ~95 mL H₂O/MeOH (95/5, v/v), add 450 µL concentrated HAc and fill to 100 mL with H₂O/MeOH (95/5, v/v). Adjust pH carefully to 6.0 with concentrated HAc. Use ultrapure 18 MΩ*cm water and LC-MS grade MeOH for H₂O/MeOH (95/5, v/v).
- **Buffer for sample loading (SPE): 0.1 M Na₂HPO₄/HAc (pH 6.0) in H₂O**
Same procedure as above, but dissolve disodium phosphate dihydrate in ultrapure water (18 MΩ*cm).
- **Ethylacetate/*n*-hexane (75/25, v/v) with 1% HAc**
Mix 750 mL ethyl acetate with 250 mL *n*-hexane, discard 10 mL (measuring cylinder) and add 10 mL HAc (LC-MS grade).

- **0.6 M KOH in MeOH/H₂O (75/25, v/v)**
Add ~20 mL MeOH/H₂O (75/25, v/v) to 1.98 g potassium hydroxide (85%) and dissolve (CAVE: exothermic reaction, work on ice). Following complete dissolution, fill up to 50 mL with MeOH/H₂O (75/25, v/v). Use ultrapure 18 MΩ*cm water and LC-MS grade MeOH for MeOH/H₂O (75/25, v/v).
- **MeOH/H₂O (50/50, v/v)**
Use LC-MS grade MeOH and ultrapure 18 MΩ*cm water.

Eluents for LC-MS/MS analysis

- **Aqueous Eluent (A)**
Add 50 mL organic eluent (B) and 950 μL HAc to 950 mL water. Use LC-MS grade HAc and ultrapure 18 MΩ*cm water.
- **Organic Eluent (B)**
Add 150 mL MeOH and 1 mL HAc to 800 mL ACN. Use LC-MS grade MeOH, ACN and HAc as well as ultrapure 18 MΩ*cm water.

Table S1: Materials and chemicals used for the described experiments together with information on suggested suppliers and article number.

	specification	supplier	article number
Materials - for cell culture			
cryo tube	2 ml	Greiner Bio-One (Solingen, Germany)	126263
conical centrifuge tube	15 ml	Sarstedt (Nümbrecht, Germany)	62.554.502
	50 ml		62.547.254
	10 ml		86.1254.001
serologic pipette	25 ml	Sarstedt (Nümbrecht, Germany)	86.1685.001
pasteur pipette		VWR (Darmstadt, Germany)	612-1701
Alternative: pasteur pipette		Carl Roth (Karlsruhe, Germany)	4518.1
dishes for thawing (adherent cells)	T25 flask, 25 cm ² (92 x 51 x 29 mm), 3 - 15 ml	Techno Plastic Products (Trasadingen, Switzerland)	90025
dishes for maintenance (adherent cells)	60.1 cm ² (ID x H, 87 cm x 21 mm)	Techno Plastic Products (Trasadingen, Switzerland)	93100
dishes for thawing (TC tested, suspension cells)	22.1 cm ² (ID x H, 60 x 15 mm), 5 ml	Sarstedt (Nümbrecht, Germany)	83.3901.500
dishes for differentiation (TC tested, suspension cells)	58.8 cm ² (ID x H, 100 x 20 mm), 13 ml	Sarstedt (Nümbrecht, Germany)	83.3902.500
cell scraper	Length 195 mm, Width of Blade 14 mm, PP	Techno Plastic Products (Trasadingen, Switzerland)	99010
1.5 ml reagent tube		Sarstedt (Nümbrecht, Germany)	72.706
2 ml reagent tube		Sarstedt (Nümbrecht, Germany)	72.691
cell freezing container		Thermo Fisher Scientific (Darmstadt, Germany)	5100-0036
Materials - for SPE and LC-MS analysis			
Autosampler vials	1.5 ml short thread vial (9-425), 32 x 11.6 mm, amber glass	IVA Analysentechnik (Meerbusch, Germany)	70911302
Inserts for autosampler vial	0.1 ml micro insert, 31 x 6 mm, 15 mm tip, clear glass	IVA Analysentechnik (Meerbusch, Germany)	70906500
Screw caps for autosampler vial	9 mm PP short thread cap, black with 6 mm hole, silicon white/PTFE red, 45° shore A	IVA Analysentechnik (Meerbusch, Germany)	71509326
Glass test tubes	70 x 10 mm	LAT (Labor und Analysetechnik) (Garbsen, Germany)	11 18 14004
Hamilton syringe 500 µL	gastight, 22/51mm/pst.3	Hamilton Germany GmbH (Gräfelfing, Germany)	1750 LTN
Hamilton syringe 2.5 mL	gastight, 22/51mm/pst.3	Hamilton Germany GmbH (Gräfelfing, Germany)	1002 LTN
Hamilton repeating dispenser		Hamilton Germany GmbH (Gräfelfing, Germany)	PB600-1
Pipette Tips (20 µL)		StarLab (Hamburg, Germany)	S1110-3000
Pipette Tips (200 µL)		StarLab (Hamburg, Germany)	S1113-1006
Pipette Tips (1000 µL)		StarLab (Hamburg, Germany)	S1111-6000
Alternative pipette tips : 20 µL		Sarstedt (Nümbrecht, Germany)	70.1114.100
300 µL		Sarstedt (Nümbrecht, Germany)	70.3040
1000 µL		Sarstedt (Nümbrecht, Germany)	70.3050
SPE cartridges	Bond Elut Certify II 200mg 3ml; 50/pk	Agilent (Waldbronn, Germany)	12102080
pH sticks	pH-Fix 5.1-7.2; 100/pk	Macherey-Nagel (Dueren, Germany)	92140
Chemicals - for cell culture			
DMEM	1000 mg/L glucose and L-glutamine, without sodium bicarbonate with sodium bicarbonate, without L-glutamine, sterile-filtered (liquid)	Sigma/Merck (Taufkirchen, Germany)	D5523
RPMI 1640	without sodium bicarbonate, with L-glutamine (powder)	Sigma/Merck (Taufkirchen, Germany)	R0883
RPMI 1640		Thermo Fisher Scientific/Life Technologies (Darmstadt, Germany)	31800089
Sodium chloride (NaCl)	99.5%, 1 kg	Grüssing GmbH (Filsum, Germany)	121221000
Potassium chloride (KCl)	≥99.5 %, 1 kg	Carl Roth (Karlsruhe, Germany)	6781.1
di-Sodium hydrogen phosphate (Na ₂ HPO ₄ , anhydrous)	≥99 %, 1 kg	Carl Roth (Karlsruhe, Germany)	P030.2
Potassium dihydrogen phosphate (KH ₂ PO ₄)	≥99 %, 1 kg	Carl Roth (Karlsruhe, Germany)	3904.1
FBS Superior	fetal bovine serum superior standardized tested for virus and mycoplasma, testedd for endotoxin	Biochrom GmbH (Berlin, Germany)	S 0615
Penicillin/Streptomycin (P/S)	5000 units Penicillin, 5 mg streptomycin/ml; 100 ml	Sigma/Merck (Taufkirchen, Germany)	P4458
L-Glutamine	200 mM, solution, sterile-filtered, BioXtra; 100 ml	Sigma/Merck (Taufkirchen, Germany)	G7513
Tris(hydroxymethyl)aminomethane (TRIS)	≥ 99.9%; 1 kg	Carl Roth (Karlsruhe, Germany)	5429.3
Trypsin from porcine pancreas	25 g/L in 0.9 % NaCl sterile-filtered, BioReagent; 100 ml	Sigma/Merck (Taufkirchen, Germany)	T4549
Ethylenediaminetetraacetic acid (EDTA)	disodium salt dihydrate, ≥ 99%, p.a, ACS; 250 g	Carl Roth (Karlsruhe, Germany)	8043.1
Dimethylsulfoxide (DMSO)	≥ 99.8%, p.a; 1 L	Carl Roth (Karlsruhe, Germany)	4720.1
Mycoplasma test	Venor GeM OneStep	Minerva biolabs (Berlin, Germany)	11-0825
Transforming growth factor-β1 (TGF-β1)	HEK293 cells derived recombinant human TGF-β1	PeproTech (Hamburg, Germany)	100-21
Calcitriol (1α,25-dihydroxy cholecalciferol)	≥ 97%; 5 mg	Cayman Chemical; local distributor biomol (Hamburg, Germany)	71820
Lipopolysaccharide (LPS) from <i>E. coli</i> (0111:B4)	purified by phenol extraction; 25 mg	Sigma/Merck (Taufkirchen, Germany)	L2630
Protease inhibitor mix	AEBBSF, 129 Aprotinin, Bestatin, E-64, Leupeptin and Pepstatin A	SERVA Electrophoresis GmbH (Heidelberg, Germany)	39102.01
Resazurin-Na-salt	p.a	SERVA Electrophoresis GmbH (Heidelberg, Germany)	34226.02
Dexamethasone	≥ 98%	Cayman Chemical; local distributor biomol (Hamburg, Germany)	11015
Indomethacin	≥99%; 1 g	Cayman Chemical; local distributor biomol (Hamburg, Germany)	70270
Celecoxib	≥ 98%	Sigma/Merck (Taufkirchen, Germany)	SML3031
tert-butyl hydroperoxide (t-BOOH)	70 wt. % in H ₂ O	Sigma/Merck (Taufkirchen, Germany)	458139
Chemicals - for SPE and LC-MS analysis			
Butylated hydroxytoluene (BHT)	≥ 99%; 1kg	Sigma/Merck (Taufkirchen, Germany)	W218405
trans-4-[4-(3-Adamantan-1-yl-ureido)-cyclohexyloxy]-benzoic acid (t-AUCB)	≥ 90%;10 mg	Cayman Chemical; local distributor biomol (Hamburg, Germany)	16568
15-LOX-1 inhibitor BLX3887	≥ 95%	Cayman Chemical; local distributor biomol (Hamburg, Germany)	27391
Acetic acid	99.7%, ACS reagent, Acros; 500 mL	Thermo Fisher Scientific (Darmstadt, Germany)	423225000
Disodium hydrogen phosphate dihydrate	≥ 99.0 %	Carl Roth (Karlsruhe, Germany)	4984.1
Ethyl acetate	HPLC grade, certified acidity < 0.00003 MEQ/g; 2.5 L	Fisher Scientific (Schwerte, Germany)	E/0906/17
n-Hexane	Rottisolv HPLC; 2.5 L	Carl Roth (Karlsruhe, Germany)	7339.1
Alternative: n-Hexane	HPLC grade; 2.5 L	Fisher Scientific (Schwerte, Germany)	H/0406/17
Glycerol	99.5%, water free; 1 L	Sigma/Merck (Taufkirchen, Germany)	G7893
Potassium Hydroxide	85%, for analytical purposes; 1 kg	Grüssing GmbH (Filsum, Germany)	120381000
Isopropanol	Optima LC/MS Grade; 2.5 L	Fisher Scientific (Schwerte, Germany)	A461-212
Methanol	Optima LC/MS Grade; 4 L	Fisher Scientific (Schwerte, Germany)	10402824
Acetonitrile	HPLC-MS Grade; 2.5 L	Fisher Scientific (Schwerte, Germany)	A/0638/17
Acetic acid	Optima LC/MS Grade; 50 mL	Fisher Scientific (Schwerte, Germany)	A113-50